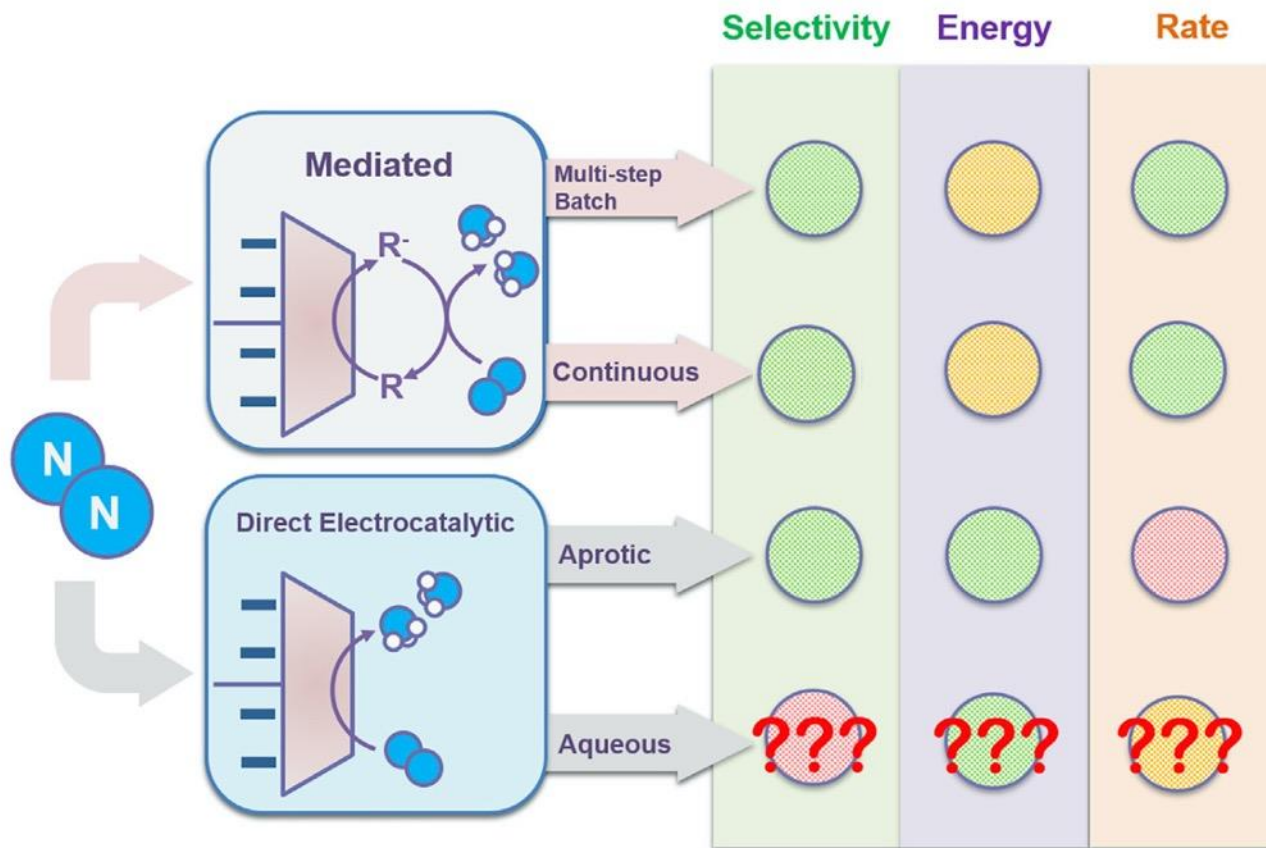


# Whither Aqueous Electro-reduction of Nitrogen to Ammonia? or



Doug MacFarlane  
**Monash University**

# Approaches to Ambient Temperature $N_2 \rightarrow NH_3$



MacFarlane et al *Joule* 2020

# ***Aqueous NRR – Current Status***

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Many reports.....*but!!!*

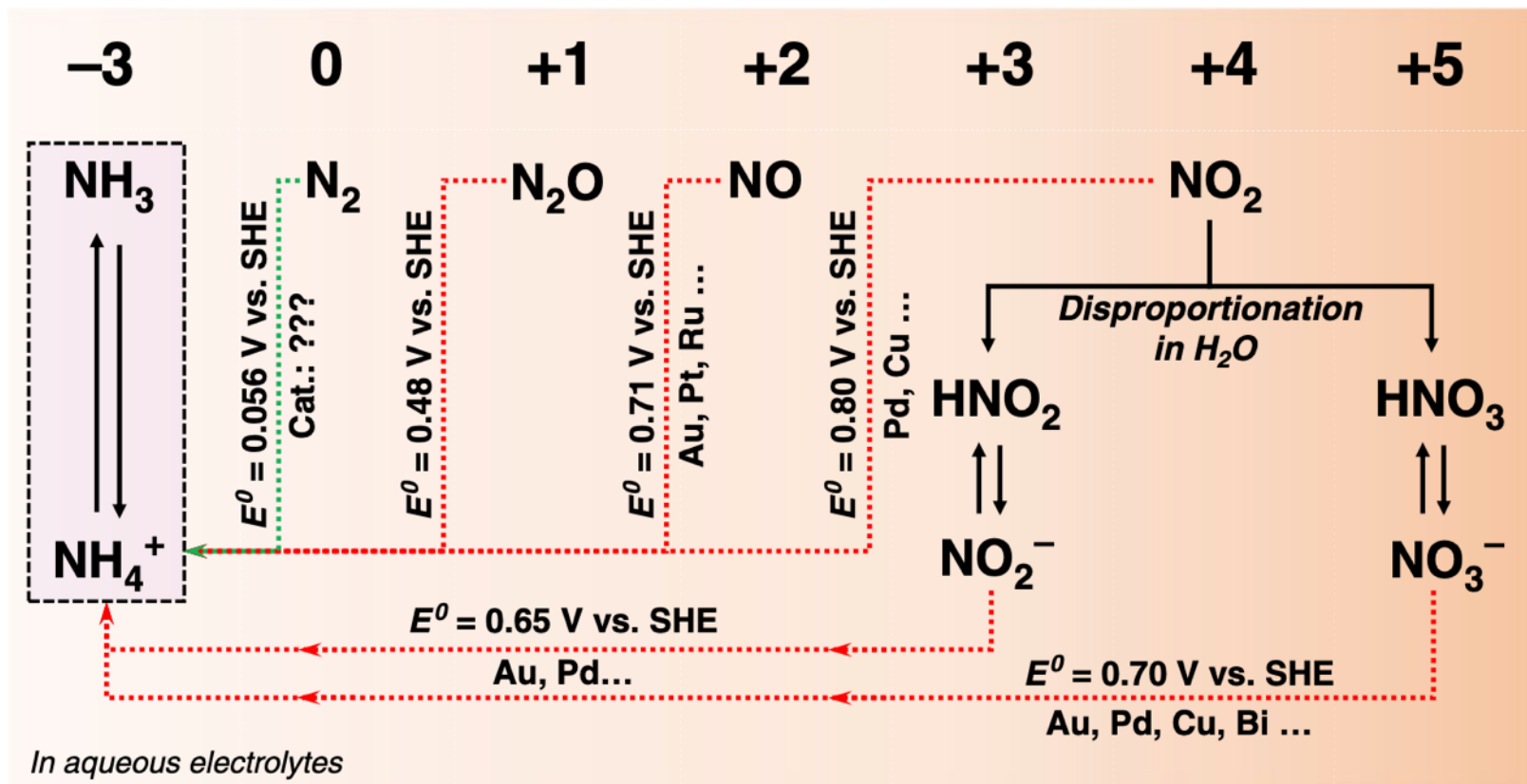
- Selectivity rarely > 20% (not practical versus Gen 2)
- Yields **so low** => prone to false positives
- Often using huge volumes of N<sub>2</sub> gas  
=> very small amounts of Nox's problematical
- Papers are now being refuted:
  - Du et al. *ACS Sust Chem. Eng.* 2019.
  - Hu et al. *ACS Energy Lett.* 2019 & Editor's Commentary
  - Has et al *Nature Cat* 2019 Matters Arising 2020
  - Choi et al *Nature Comms* 2020

# The “Richness” of Nitrogen Chemistry!

Oxidation  
state

Gaseous  
forms

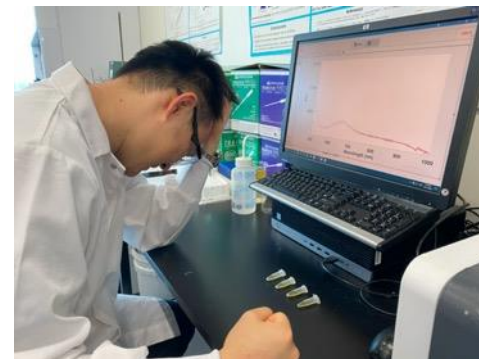
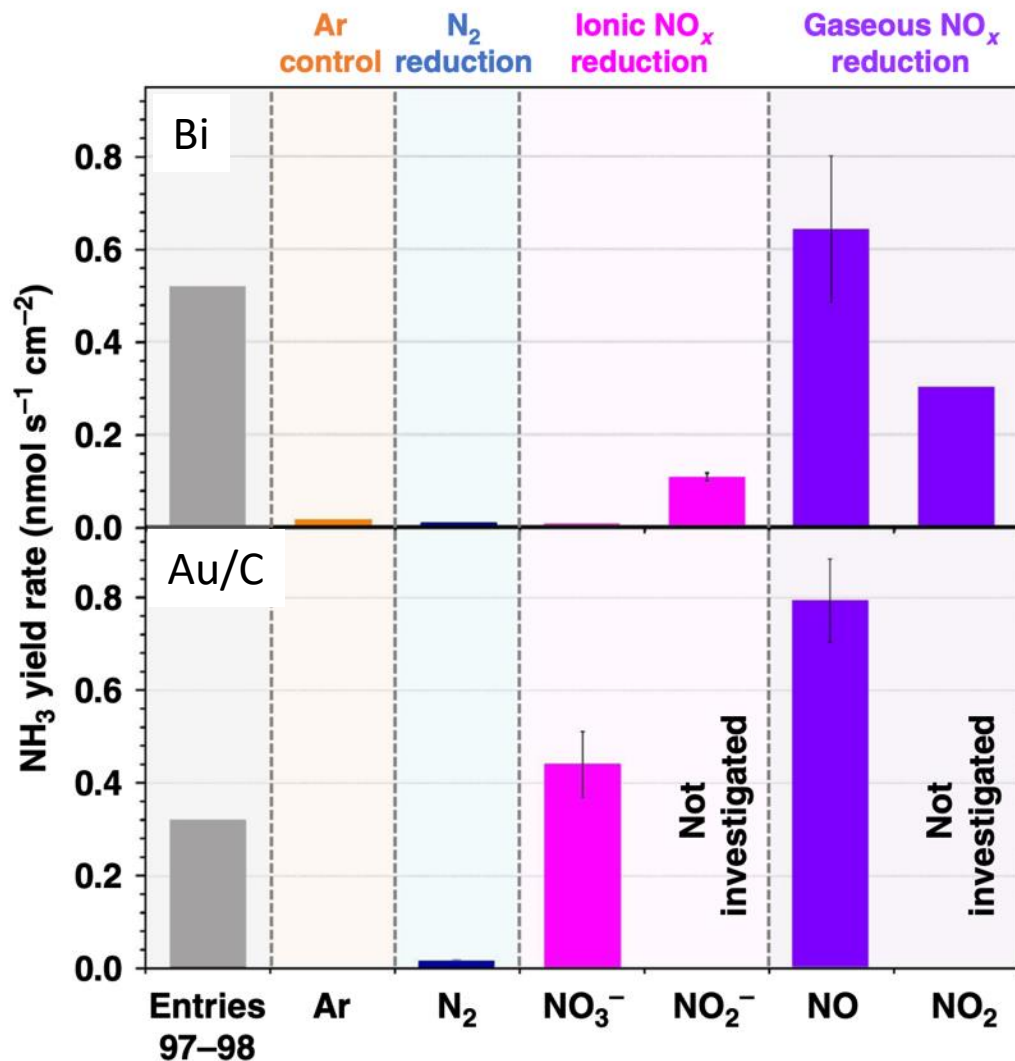
Ionic  
forms



Choi et al *Nat Comm* 2020

Choi et al *ACS Energy Letters* 2020

# => *Frustratingly False Positives!*



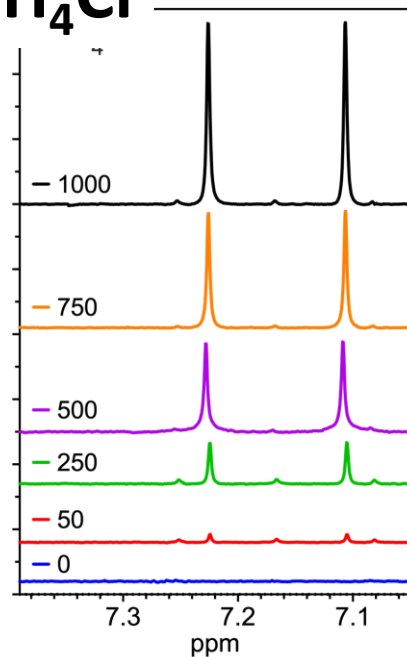
*A very frustrated Jay Choi*

# $^{15}\text{N}_2$ studies important to prove NRR.....

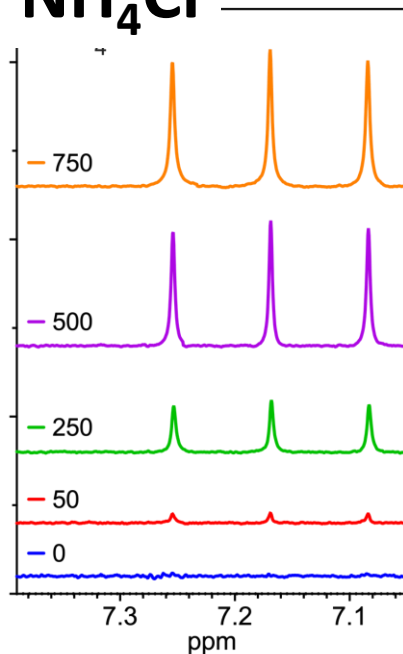
.....but not foolproof

The difference is clear in the  $^1\text{H}$  NMR:

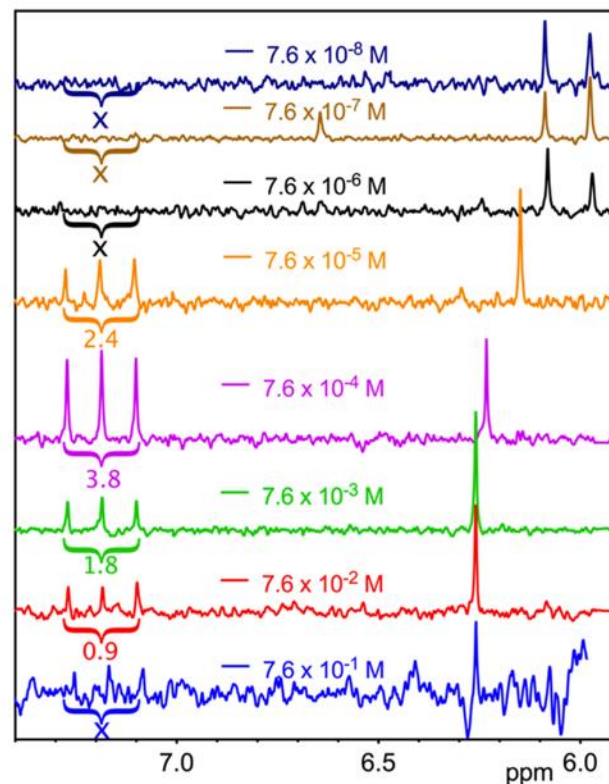
$^{15}\text{NH}_4\text{Cl}$



$^{14}\text{NH}_4\text{Cl}$



But it depends critically on conditions  
Eg pH:



# A Traffic Light Analysis of Literature

No.	Year	Material	N-based material	Electrolyte	<i>E</i> (V vs. RHE)	FE (%)	Yield rate (nmol s <sup>-1</sup> cm <sup>-2</sup> )	Quantitative <sup>15</sup> N <sub>2</sub> test	NO <sub>x</sub> control	Ref.
19	2018	Fe/Fe <sub>3</sub> O <sub>4</sub>	N	0.1 M PBS	-0.3	8.3	0.003	N	N	19
20	2018	Ag	N	0.1 M HCl	-0.6	4.8	0.046	N	N	20
21	2018	C	N	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.3	6.9	0.26	N	N	21
22	2018	NbO <sub>2</sub>	N	0.05 M H <sub>2</sub> SO <sub>4</sub>	-0.65	20	0.19	N	N	22
23	2018	Y <sub>2</sub> O <sub>3</sub>	N	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.9	2.5	0.11	N	N	23
24	2018	d-MoS <sub>2</sub>	Y	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.4	8.3	0.19	N	N	24
25	2018	Au/CN <sub>x</sub>	Y	0.1 M HCl	-0.2	12	0.038	N	N	25
26	2018	Au/C <sub>3</sub> N <sub>4</sub>	Y	0.005 M H <sub>2</sub> SO <sub>4</sub>	-0.1	11	0.031	N	N	26
27	2018	Fe/N-CNT	Y	0.1 M KOH	-0.2	9.3	0.29	N	N	27
28	2018	B-TiO <sub>2</sub>	N	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.8	3.4	0.024	N	N	28
29	2018	B <sub>4</sub> C	N	0.1 M HCl	-0.75	16	0.043	N	N	29
30	2018	Au	N	0.1 M HCl	-0.2	6.1	0.25	N	N	30
31	2018	Mn <sub>3</sub> O <sub>4</sub>	N	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.8	3.0	0.038	N	N	31
32	2018	CN <sub>x</sub>	N	0.1 M HCl	-0.2	1.5	1.0	N	N	32
33	2018	SnO <sub>2</sub>	N	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.8	1.2	0.15	N	N	33
34	2018	TiO <sub>2</sub>	N	0.1 M Na <sub>2</sub> SO <sub>4</sub>	-0.7	2.5	0.092	N	N	34
35	2018	VN	N	0.05 M H <sub>2</sub> SO <sub>4</sub>	-0.2	6.5	0.50	N	N	35



## Conclusion:

### ➤ Gas purification and analysis vital:

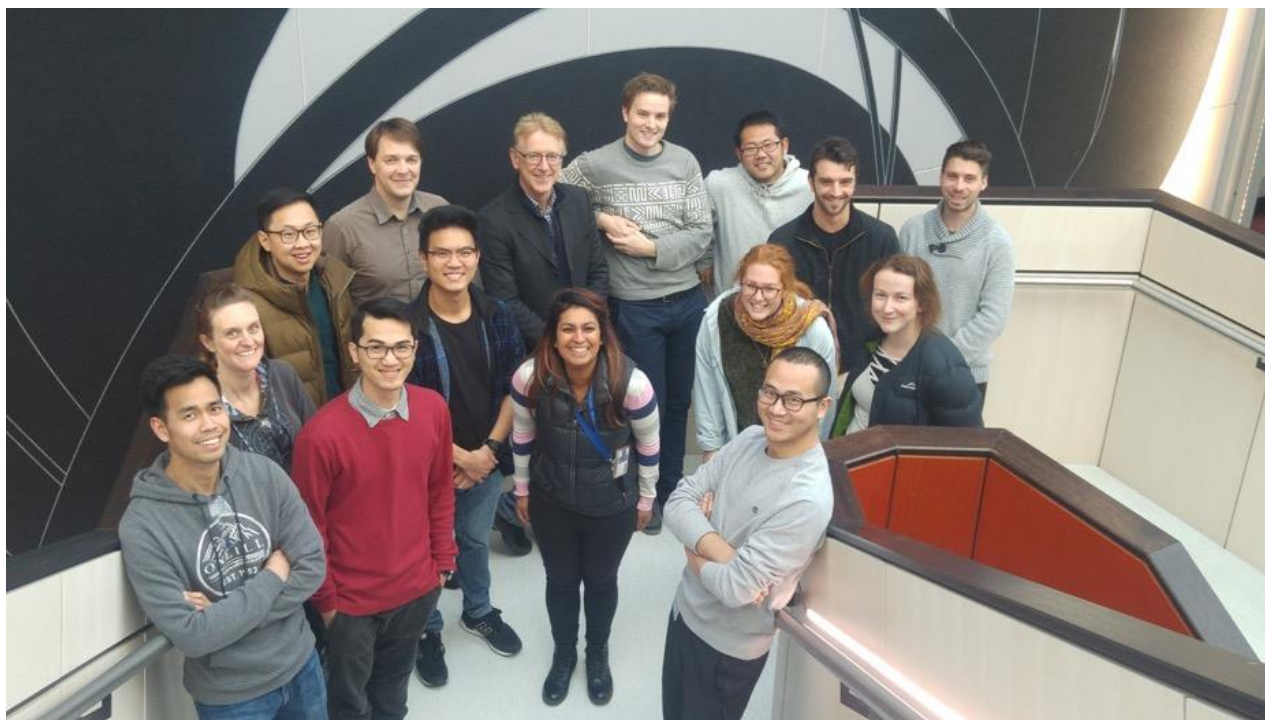
- H<sub>2</sub>O (in aprotics)
- Nitrate/nitrite
- NO<sub>x</sub> (all of them)
- NH<sub>3</sub>

### ➤ See latest protocol commentary in Choi et al *Nat Comm* 2020





## The Hydrogen/Ammonia Team



Australian Government  
Australian Research Council



Australian Government  
Australian Renewable  
Energy Agency

**ARENA**